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CHAPTER 4 TEST: Atoms, Atomic Theory and Atomic Structure Matching. A. Bohr B. Democritus C. Rutherford D. Dalton E. Thomson F. Schrodinger
____ 1. Greek thinker; called nature's basic particle an atom, based on the Greek word "atomos" which means "indivisible". Did not have evidence that atoms existed. ____ 2. <http://www.brooklyn.k12.oh.us/userfiles/75/Classes/4711/NON-HONORS%20Chapter%204%20Test%202010.pdf>.

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Chapter 4: Atom (Test Answers) STUDY. Flashcards. Learn. Write. Spell. Test. PLAY. Match. Gravity. Created by. chxistina. Key Concepts: Terms in this set (33) Dalton's atomic theory was accepted because a. Dalton showed how new molecules are formed. b. Dalton invented the electron microscope. c. Democritus said that it was correct.

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Section 4.2 Structure of the Nuclear Atom One change to Dalton's atomic theory is that atoms are divisible into subatomic particles: Electrons, protons, and neutrons are examples of these fundamental particles There are many other types of particles, but we will study these three

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Chapter 4: The Structure of the Atom

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Atomic Structure (Chapter Test A) STUDY. Flashcards. Learn. Write. Spell. Test. PLAY. Match. Gravity. Created by. emily_reidt. Terms in this set (21) the total number of protons and neutrons in the nucleus of an atom. mass number. the weighted average mass of the atoms in a naturally occurring sample of an element.

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